Week 3 Lesson 2

Desalination

These activities should take about 1 ½ lessons.

Aim: To show how fresh water can be obtained from salt water and how this process happens in the natural world.

Keywords: aquifer, condensation, distillation, evaporation, hypotonic, hypertonic, osmosis, precipitation, proteins, semi-permeable membrane, solution, water cycle

Starter activities

1. Homework recap

Students use the homework which was set in preparation for the lesson. They can read their partners' work and then tell the class about it.

2. How does nature provide us with fresh water?

Sketch and label the diagram of the water cycle (see PowerPoint).

Discuss how snow melt in the mountains during warmer months provides many communities downstream with a constant supply of fresh water during the summer when there is little rainfall. Global warming is causing snow fields and glaciers to melt. In time, this source of fresh water may diminish or disappear completely.

The video students watched for homework mentioned aquifers. Explain what these are and how bore holes have been sunk into underground water sources. Often, water is being extracted faster than it is replenished by seepage through rocks from the ground above. Saline water can move in to replace the fresh water removed in this way.

And on a smaller scale, how does water move in and out of cells?

Main activities

1. World's population without safe drinking water

The following activity illustrates the difficulty for many people across the globe of having access to safe, clean drinking water. It also provides an opportunity to develop students' mathematical skills. Use the PowerPoint, 'World's population and water', and invite students up to write their answers on the interactive whiteboard. Students will each need access to a calculator.

Differentiation

Students who need some support could work out the population in whole numbers whereas others could work out the population in standard form.

Assessment opportunity

The activity provides an opportunity to assess the confidence of students when making estimates, rounding numbers, calculating percentages and working in standard form.

2. The process of desalination

Freshwater can be produced by taking salt out of seawater, desalination. It uses physical processes, evaporation, condensation and osmosis. Explore different methods of desalination in these two practical activities.

Practical 1: Osmosis demonstration

N.B. This demonstration needs to be prepared well in advance of the lesson. Remove the shell from 2 chicken eggs by dissolving the shell in acetic acid. This will take 2 – 3 days.

Preparation steps

a. Find the mass of each egg.

b. Place one egg in a hypotonic solution and the other in a strongly hypertonic solution.

c. Leave for at least 8 hours.

d. Find the mass of each egg.

Explain the egg is a single cell. Beneath the shell is a semi-permeable membrane. The egg contains water, salts and large molecules such as proteins. Water moves across the membrane, from an area of high concentration to an area of low concentration. This is known as osmosis.

Explain that there are two methods of desalination by osmosis that are used- reverse and forward osmosis.

Reverse osmosis

Pressure is applied to push water through a semi-permeable membrane. The membrane is designed to prevent larger solutes from passing through.

Problems: It requires a lot of energy and is expensive. The membranes become clogged with bacteria and they deteriorate when chlorine is used to kill the bacteria.

Forward osmosis

A semi-permeable membrane separates the saline water from a highly concentrated solution of ammonia salts. Water molecules move across the membrane into the highly concentrated solution.

The ammonia solution is heated to evaporate the ammonia salts which are collected and reused.

This is a relatively new technology and needs funding for further research and development.

Differentiation

Ask students to draw annotated diagrams to describe (or explain) one of the types of osmosis used in desalination.

Extension

Students could annotate diagrams to compare the advantages and disadvantages of the two methods.

To see the latest developments watch the following clips.

The first shows how proteins are being incorporated into membranes:

www.youtube.com/watch?v=1zcEYYRtSnE

A short clip showing how graphene is used to make a filter:

www.youtube.com/watch?v=k5Tjy_90WBU

Practical 2: Distillation demonstration

Set up a demonstration (or student practical) to show how pure water can be distilled from salty water.

Explain that desalination by distillation requires a great deal of energy to heat the water. The idea of specific heat capacity of water could be explored at this point.

This illustrates that desalination by distillation requires a great deal of energy and is expensive. The cost of building and maintaining these facilities is also very high.

Plenary activity

1. What's the question?

Put the answers on the board – students write the question

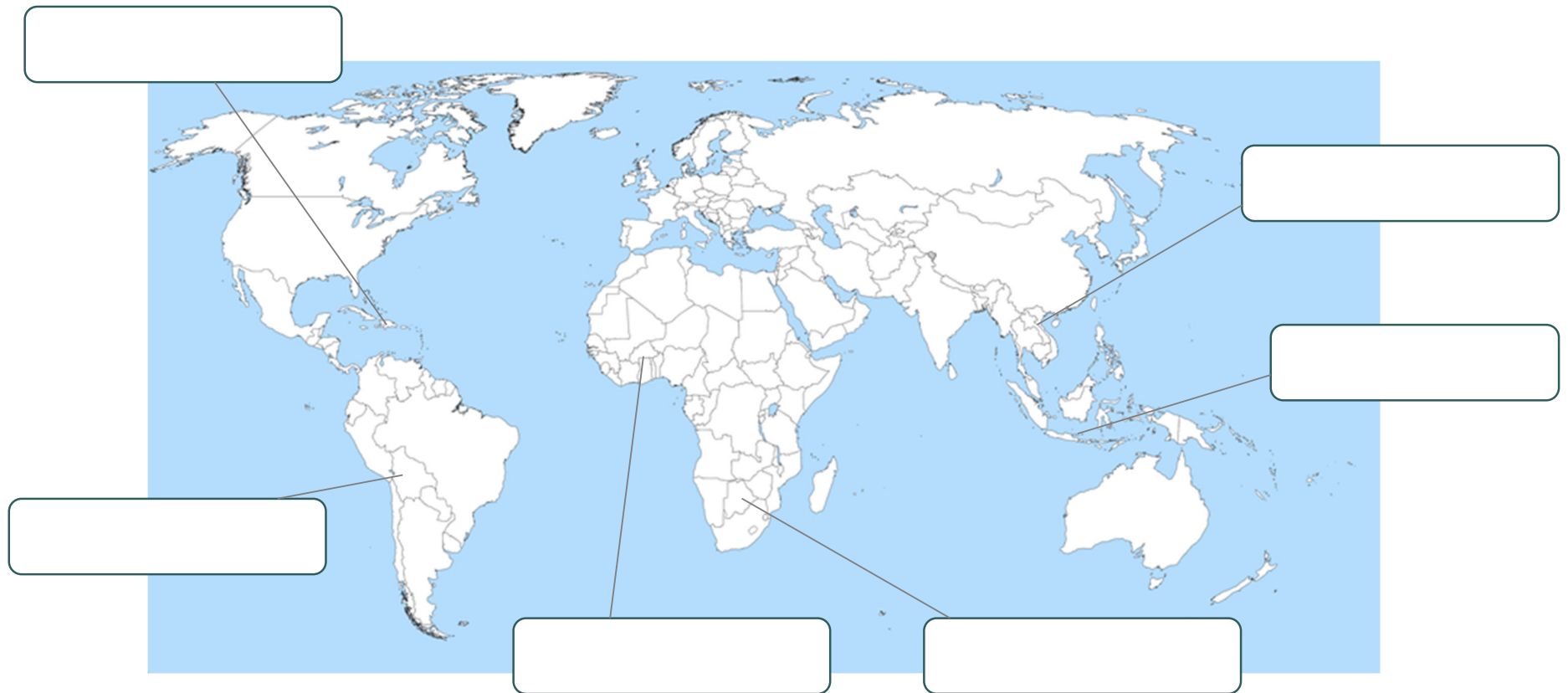|	
a.	The water cycle.
b.	A natural store of underground water.
c.	By distillation.
d.	It allows water to pass into and out of cells but not large molecules such as proteins.
e.	Osmosis.
f.	These are all disadvantages of desalination by distillation.

Starter 2

How does nature provide us with fresh water? — Answers
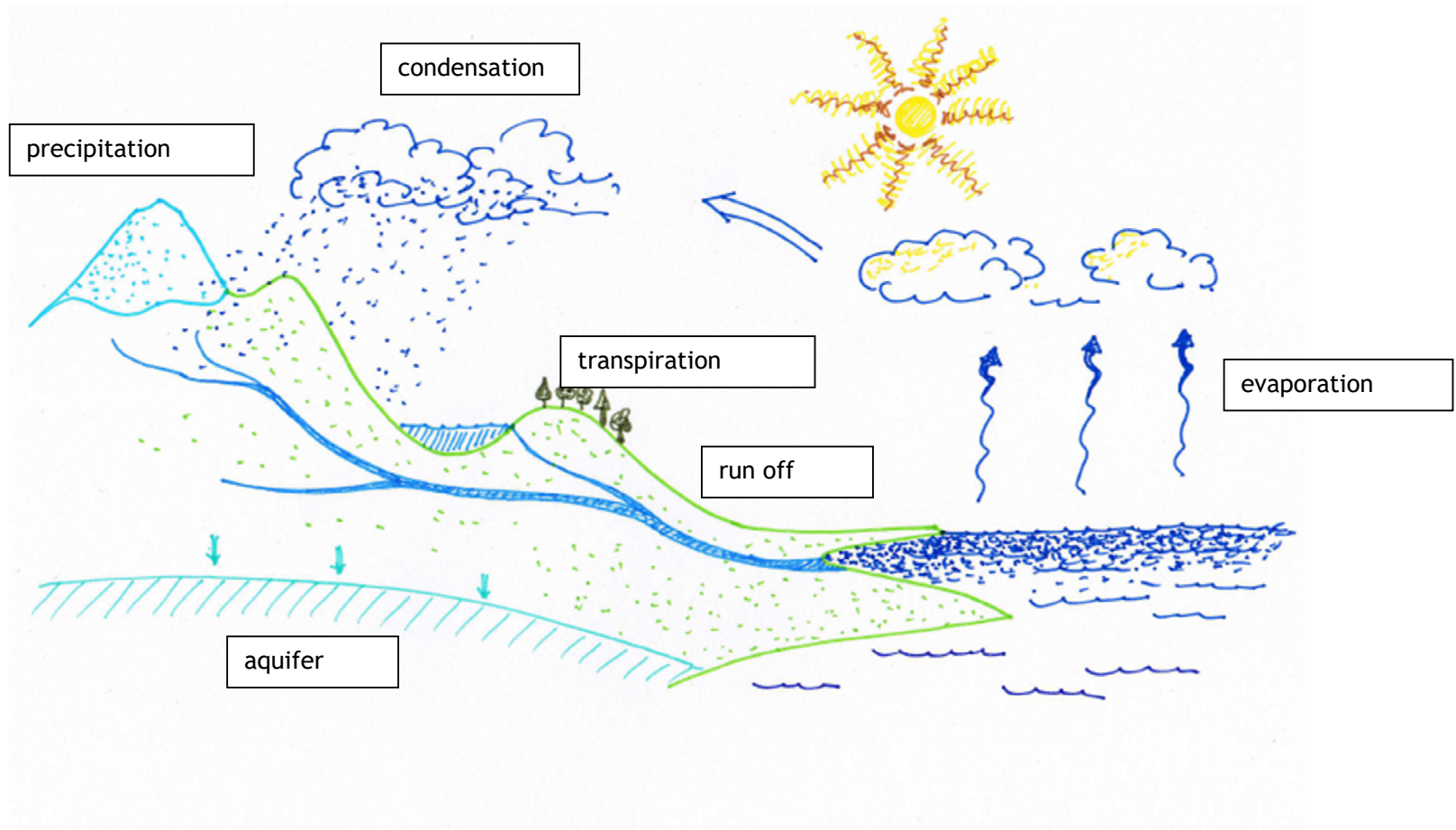
Main 1

World's population without safe drinking water

Main 2

World's population without safe drinking water — PowerPoint

Slide 1 (Teaching notes)

Teaching notes

This activity is differentiated so that some students could work out the population of each country in whole numbers whereas others could work out the population in standard form.

Differentiation

Ask them to note down the name of the country and its population as a whole number or in standard form and then show the last slide to see if they can name each country from the information they have got. This should be slightly more difficult for the students who have got the population written down in standard form.

Slide 2

World's population and water

Slide 3

Add your estimates to the table.

Country	Total population (millions)	Population without safe drinking water (millions)	Estimation of percentage of population without safe drinking water
Bolivia	10.91	6.89	
Botswana	2.02	1.82	
Burkina Faso	17.05	7.16	
Dominican Republic	10.48	6.82	
Indonesia	249.90	187.50	
Laos	6.77	2.98	

Slide 4

Country	Total population (millions)	Population without safe drinking water (millions)	Percentage of population without safe drinking water (to nearest whole number) Calculation	%
Bolivia	10.91	6.89	(6.89 million ÷ 10.91 million) x 100	?
Botswana	2.02	1.82		?
Burkina Faso	17.05	7.16		?
Dominican Republic	10.48	6.82		?
Indonesia	249.90	187.50		?
Laos	6.77	2.98		?

Why is it helpful to use percentage values rather than the unprocessed population data?

Enables you to make comparisons between countries easily.

Slide 5

Write down the country and its population from the third or fourth column.

Country	Total population (millions)	Population as a whole number	Population as a whole number rounded to the nearest million	Extension: Population in standard form to 2 dp
Bolivia	10.91			
Botswana	2.02			
Burkina Faso	17.05			
Dominican Republic	10.48			
Indonesia	249.90			
Laos	6.77			

Slide 6

Can you use your notes to identify each country by its population? Click a box to check your answer. Label the countries on your paper version of the map.

10 million, 7 million, 250 million, 11 million, 17 million, 2 million

Slide 7

Write down the country and its population from the third or fourth column.

Country	Total population (millions)	Population as a whole number	Population as a whole number rounded to the nearest million	Extension: Population in Standard form to 2 decimal places
Bolivia	10.91	10 910 000	11 million	1.09×10^7
Botswana	2.02	2 020 000	2 million	2.02×10^6
Burkina Faso	17.05	17 050 000	17 million	1.71×10^7
Dominican Republic	10.48	10 480 000	10 million	1.05×10^7
Indonesia	249.90	249 900 000	250 million	2.50×10^8
Laos	6.77	6 770 000	7 million	6.77×10^6

Slide 8

How can scientists help solve the problem of a lack of safe drinking water using processes found in nature?

The processes, evaporation/condensation in the water cycle and osmosis, are found in nature, and are used by people to produce fresh water from salt water in a process known as desalination.

Desalination was mentioned in the video you watched for homework.